

Balancing Equations Gizmo Key

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Balancing Chemical Equations Balance and classify five types of chemical reactions: synthesis, decomposition, single replacement, double replacement, and combustion. While balancing the reactions, the number of atoms on each side is presented as visual, histogram, and numerical data. 5 Minute Preview

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Balance and classify five types of chemical reactions: synthesis, decomposition, single replacement, double replacement, and combustion. While balancing the reactions, the number of atoms on each side is presented as visual, histogram, and numerical data.

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The equation is balanced when there are equal numbers of each type of atom represented on each side of the equation. In the Gizmo, use the up and down arrows to adjust the numbers of hydrogen, oxygen, and water molecules until the equation is balanced. When you are done, turn on

[BalancingChemEquationsSE.docx](#)

This process is summarized by a chemical equation. In the Balancing Chemical Equations Gizmo, look at the floating molecules below the initial reaction: $H_2 + O_2 \Rightarrow H_2O$. 1. How many atoms are in a hydrogen molecule (H_2)? 2 2. How many atoms are in an oxygen molecule (O_2)? 2 3. How many hydrogen and oxygen atoms are in a water molecule (H_2O)? H: 2 O: 1 4. In general, what does a subscript (such as the "2" in H_2)

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Balancing Chemical Equations Answer Key. Balancing Chemical Equations -Answer Key Balance the equations below: 1) $1 N_2 + 3 H_2 \rightarrow 2 NH_3$ 2) $2 KClO_3 \rightarrow 2 KCl + 3 O_2$ 3) $2 NaCl + 1 F_2 \rightarrow 2 NaF + 1 Cl_2$ 4) $2 H_2 + 1 O_2 \rightarrow 2 H_2O$ 5) $1 Pb(OH)_2 + 2 HCl \rightarrow 2 H_2O + 1 PbCl_2$ 6) $2 AlBr_3 + 3 K_2SO_4 \rightarrow 6 KBr + 1 Al_2(SO_4)_3$

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Therefore, a balanced chemical equation will show the same number of each type of atom on each side of the equation. To set up an equation in the Chemical Equations Gizmo, type the chemical...

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Balancing Chemical Equations Answer Key Vocabulary: coefficient, combination, compound, decomposition, double replacement, element, molecule, product, reactant, single replacement, subscript Prior Knowledge Questions (Do these BEFORE using the Gizmo.) [Note: The purpose of these questions is to activate prior knowledge and get students thinking.

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Student Exploration For Gizmo Answer Key Chemical Equations

HBr 4 NH₄OH - KAl (SO₄)₂ · 12H₂O - Al (OH)₃ (NH₄)₂SO₄ - KOH 12 H₂O When you balance the chemical equation, it's always a good idea, to check the final equation, to make sure it works. Check: Reduce the numbers of each type of atom. The total number of atoms in the balanced equation will be the same on both sides of the equation.

Student exploration balancing chemical equations gizmo ...

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When balancing equations, remember chemical reactions must satisfy conservation of mass. Check your work to make certain you have the same number and type of atoms on the reactants side as on the products side. A coefficient (number in front of a chemical) is multiplied by all the atoms in that chemical.

Balancing Equations Chemistry Test Questions

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Chemical Equations Answers

Balancing Equations: Answers to Practice Problems 1. Balanced equations. (Coefficients equal to one (1) do not need to be shown in your answers).

Balancing Equations: Practice Problems

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