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In Example 12.2.1 and Example 12.2.2, the identity of the limiting reactant has been apparent: $[\text{Au}(\text{CN})_2]^-$, LaCl_3 , ethanol, and para-nitrophenol. When the limiting reactant is not apparent, we can determine which reactant is limiting by comparing the molar amounts of the reactants with their coefficients in the balanced chemical equation, just as we did in Section 11.4 .

~~Chapter 12.2: Stoichiometry of Reactions in Solution ...~~

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Chapter 12 Test Stoichiometry Pearson Chapter 12 Review 2 Multiple Choice Identify the letter of the choice that best Page 13/30. Access Free Chemistry Chapter 12 Test B Stoichiometry completes the statement or answers the question. 1. binds the atoms together is called a(n) A mutual electrical attraction

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