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1 Answers

1) Zinc and lead (II)

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nitrate react to form zinc nitrate and lead.

2) Aluminum bromide and chlorine gas react to form aluminum chloride and bromine gas. 3) Sodium phosphate and calcium chloride react to form calcium phosphate and sodium chloride. 4) Potassium metal and chlorine gas combine

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to form potassium
chloride.

Chapter 7 Worksheet

#1 Balancing

Chemical Equations

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1. What is the
difference between
cellular respiration
and breathing?

Cellular respiration -
production of ATP
energy from food and

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oxygen. Breathing –
inhalation of oxygen
and exhalation of
carbon dioxide 2.

Briefly explain
oxidation and
reduction reactions.

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1. Given the following

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reactions, draw reasonable mechanisms that describe the chemical transformation that has taken place. To the right, list the name of the elementary mechanistic step used in the transformation.

+ HSO NaCl NaBH₃ .

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Solved: Chapter 7

Worksheet 1. Given

The Following Reactio

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Answers Chapter 7

Worksheet: Answers

Practice 1: 45

Practice 2: a) Practice

3: 0.6 Practice 4: 0.9

Practice 5: 5 6

Practice 6: 5 6 1 13 1

26 b) c) 1 2 d) 1 10

Practice 7: Event: 0

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Heads 3 Heads 4

Heads Probability: 1

16 4 16 6 16 4 16 1

16 Practice 8: a)

Odds for are 1:2 b)

Odds against are 2:1

c) Odds on are 2:1

Practice 9: a) 0.52

Practice 10: a)

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1. Given the following reactions, draw

reasonable

mechanisms that

describe the chemical transformation that

has taken place. To

the right, list the name

of the elementary

mechanistic step used

in the transformation

$\text{NO} + \text{NO} \xrightarrow{\text{H}_2\text{SO}_4}$

$\text{HSO}_3^- + \text{NO}_2^- \xrightarrow{\text{HSO}_4^-}$

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H₂SO₄ + NaCl → NaCl + H₂O

Answers

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The Following Reactio

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of way. 3. yielding. 4.
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green light. 6. fresh
green light. 7. joining.

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#1 Balancing

Chemical Equations

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