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What is a chemical Equation Class 10? Chemical Equation:  
Representation of chemical reaction using symbols and formulae of the substances is called Chemical Equation. Example:  $A + B \rightarrow C + D$   
In this equation, A and B are called reactants and C and D are called the products. The arrow shows the direction of the chemical reaction.

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~~Exothermic~~ The amount of heat evolved is so large that the metal is... The displacement reaction of iron (III) oxide with aluminium is known as thermite reaction.

## ~~Chemical Reactions and Equations Class 10 Important ...~~

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~~CBSE Class 10 Chapter 1 Science Notes: Chemical Reactions ...~~

**EXOTHERMIC REACTIONS** • An exothermic reaction is a chemical reaction that releases energy by light or heat. It is the opposite of an endothermic reaction. • Expressed in a chemical equation: reactants → products + energy For example 1)  $4\text{Fe}(s) + 3\text{O}_2(g) \rightarrow 2\text{Fe}_2\text{O}_3(s)$  2) Making ice cubes 3) Mixing sodium sulfite and  $\text{SO}_2$ .

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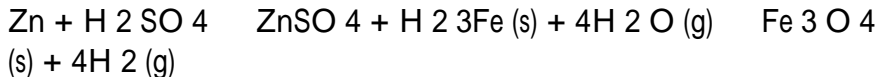
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**Chemical Equation: A mathematical statement which represents the chemical reaction. Types of Chemical Reactions: Combustion reaction – The reaction of a combustible material with an oxidiser to give an oxidised product. Decomposition reaction – The reaction in which a single component breaks down into multiple products.**

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**Balanced chemical equation** The chemical equation that shows the chemical reaction needs to be balanced. A balanced chemical equation occurs when the number of the atoms involved in the reactants side is equal to the number of atoms in the products side.



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