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Properties Matter –

Anything that has
MASS and

VOLUME. ATOMS

and MOLECULES

the basic building
blocks of matter.

ATOM-- the
smallest unit of an
element that still
maintains the
properties that
element.

ELEMENT – a pure

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Substance made up
of only one type of
atom.

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the flow of a

compressed gas

must be controlled

for practical and

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Reviewing

Vocabulary Match

the definition in

Column A with the

term in Column B.

Column A Column B

1. The set of frequencies of the electromagnetic waves emitted by the atoms of an element
2. The

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minimum amount of energy that can be lost or gained by ...

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q... the study of
energy changes

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produced or absor...

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Change. extensive
property. intensive
property. extensive
properties

(examples)

intensive property

(examples) a

property that

depends on the

amount of matter in

a sample. a

property that

depends on the type

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of matter in a
sample not.... mass,
volume, length,
weight.

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number of moles of
known substance C

H C H C H O O O

CO CO CO H O H O

H O number of

moles of unknown
substance O CO H

O C H CO H O C H

O H O C H O CO

use mole ratio 5

mol O 2 1 mol C 3

H 8 3 mol CO 2 1

mol C 3 H 8 4 mol

H 2 O 1 mol C 3 H

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8.1 mol C₃H₈
5 mol O₂

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A “chemical
change” is a change
that produces
matter with a
different

composition than
the original matter.

Chemical Change A

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Change in which one or more substances are converted into different substances. Heat and light are often evidence of a chemical change. Properties of Compounds Quite different properties than their component elements.

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