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Chapter Test B, continued 17. A substance combines with oxygen, releasing a large amount of energy as heat and light, in a(n) . 18. The decomposition of a substance by an electric current is called . 19. A(n) orders the elements by the ease

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with which they undergo certain chemical reactions. 20.

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Modern Chemistry 141 Chapter Test Name Class Date

Chapter Test B, continued 18. When a strong acid is titrated with a weak base, the pH of the solution at the equivalence

point is than 7. 19. A is a highly purified solid used to check

the concentration of a standard solution. 20. A 1 M solution of NaOH will have a pH that is

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Chapter Test A, continued _____ 19. In the figure on the

previous page, the pH at the equivalence point a. is equal to

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7.0. b. is greater than 7.0. c. is less than 7.0. d. cannot be determined from the data given. _____ 20. In the figure on the previous page, the volume of titration standard

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Modern Chemistry 105 Chapter Test Name Class Date

Chapter Test A, continued Use this figure to answer questions 7 and 8. _____ 7. A solution containing 35 g of Li_2SO_4 dissolved in 100 g of water is heated from 10°C to 90°C . According to information in the figure, this temperature change would result in a. an additional 5 g of Li_2SO_4 in ...

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...

18. Dalton's atomic theory agreed with the modern atomic theory EXCEPT for the statement that all atoms of the same element have the same Complete the following table to compare the types of subatomic particles. 19. 20. 21. 10 Particle Proton Neutron Electron CHAPTER 3 TEST Mass number Relative charge Location MODERN CHEMISTRY

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Modern Chemistry Chapter 12 Test Modern Chemistry Chapter 12. Capillary Action. Fluid. Surface Tension. Vaporization. the attraction of the surface of a liquid to the surface of a solid. a substance that can flow and therefore take the shape of its container. a force that tends to pull adjacent parts of a liquid's surface together.

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